Direct evidence of He-induced excitation process of H atoms in cooled laser plasma

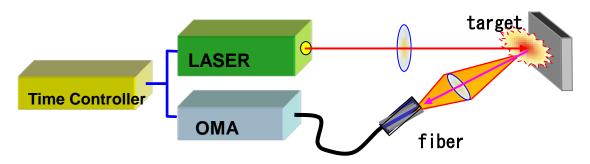
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Laser Induce Plasma Spectroscopy

Mechanism



Advantage

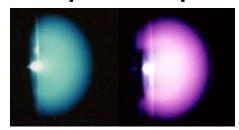
- Metal and non metal
- Without pre-treatment
- In-situ analysis
- Micro-area analysis

1 atm plasma



(Cremers & Radziemski's Group, 1981)

Low pressure plasma

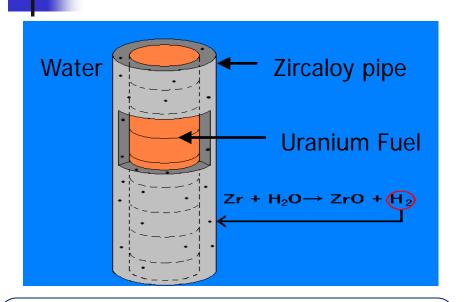


He gas plasma



Kagawa & Kurniawan's Group

Practical Application of LIBS for Hydrogen Analysis In Nuclear Power Station



Zircaloy pipe is an important material used in a light water nuclear power station to contain uranium fuel

Problem:

Hydrogen accumulation in the pipe wall degrades the pipe's strength

Anticipation:
Periodical Inspection

Now in use Inspection Method:

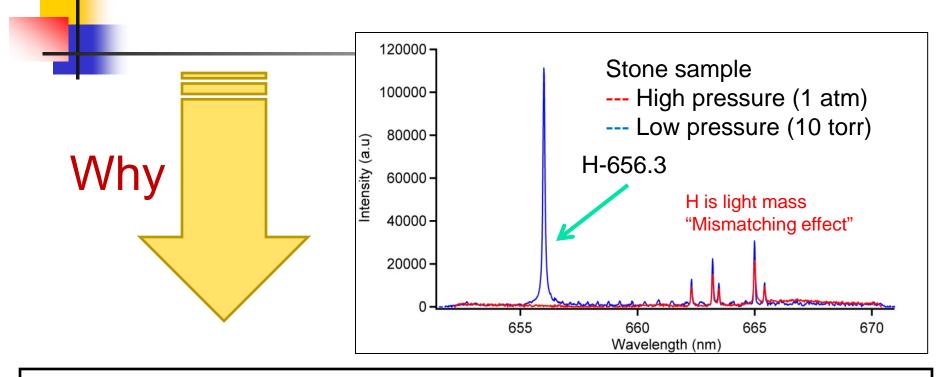
Gas Chromatography

(sample and time consuming, hand touched radioactive and not in situ analysis)

New Method:

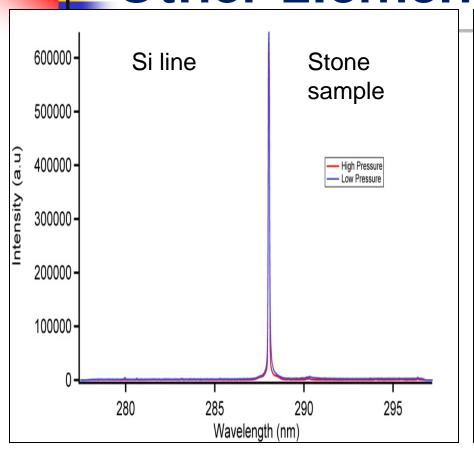
Laser Induced Plasma Spectroscopy (LIPS)

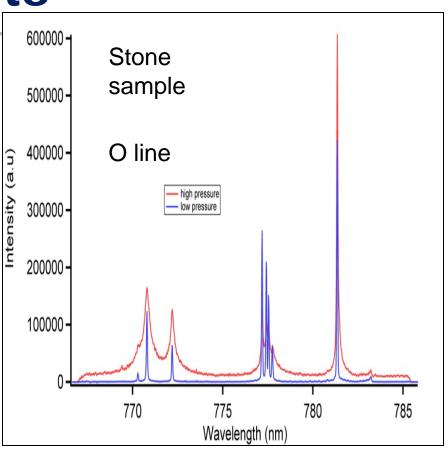
Hydrogen Analysis Using Laser Plasma Method has not been Carried Out Before



- Stark broadening effect happens when plasma is produced at 1 atm
- Emission efficiency is very low for H atom at 1 atm due to the "mismatching effect"
- Disturbance of H₂O

No Mismatching Effect for Other Elements

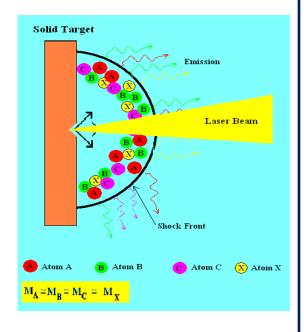




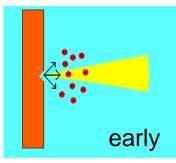
- Red line High Pressure (1 atm)
- Blue line Low Pressure (10 torr)

Concept of Mismatching Effect

Ideal Shockwave Plasma Formation



Hydrogen Atom Case



M_H=1

Hydrogen atoms gush out faster

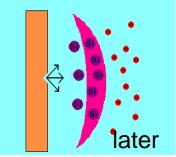
M_{Si}=28

Shock-wave generation

Mismatching Effect

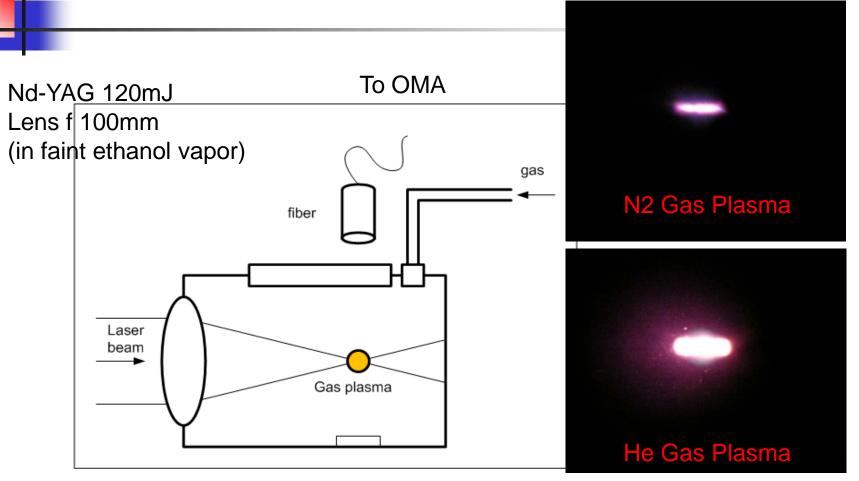


Time difference between gushing out of H atoms and shock-wave generation

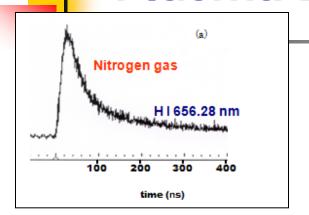


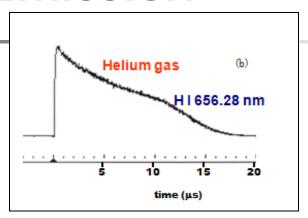
Mismatching Effect will significant at higher pressure, or at low power density, because shockwave generation is somewhat delayed

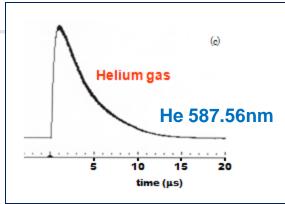
Gas Plasma Induced by Focusing Nd-YAG laser in the Gas

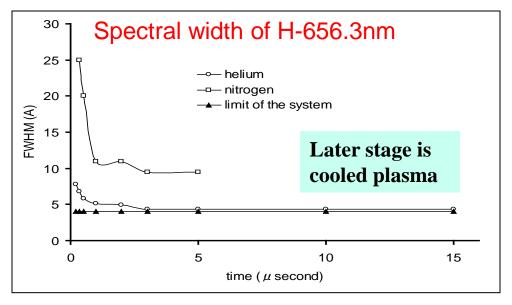


Time Profile of Laser Induced Gas Plasma Emission









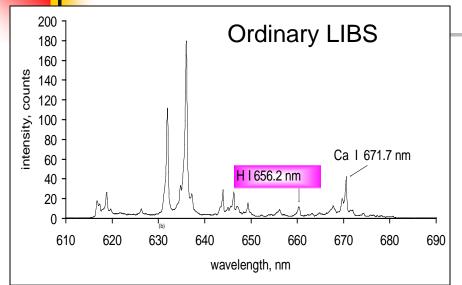
He as host gas:

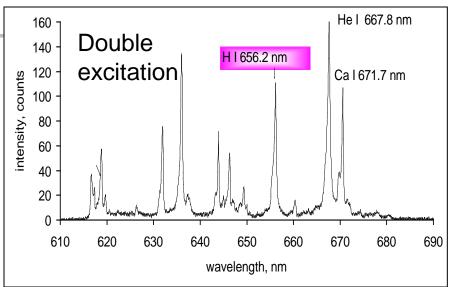
- **●Long Life Emission**
- Very narrow spectral width at later stage

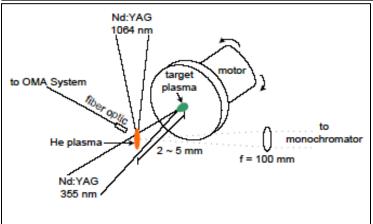
N₂ as host gas:

- **Short Life Emission**
- Rather broad spectral width

Double Excitation Scheme for Handlysis under He 1 Atm



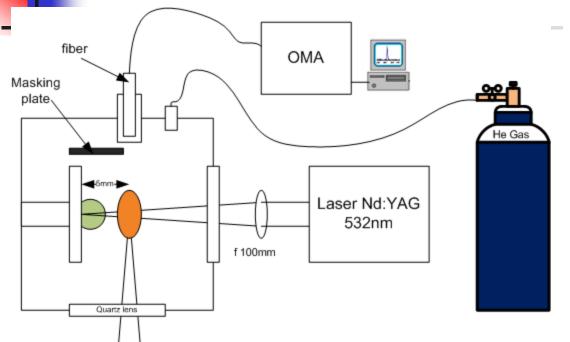




Sample: Slide glass

Hydrogen Emission can be detected with high efficiency and very narrow spectral width

Experiment Setup for Proving Excitation in Cooled Plasma



f 75mm

Laser

Nd:YAG

1064nm

Stone Sample

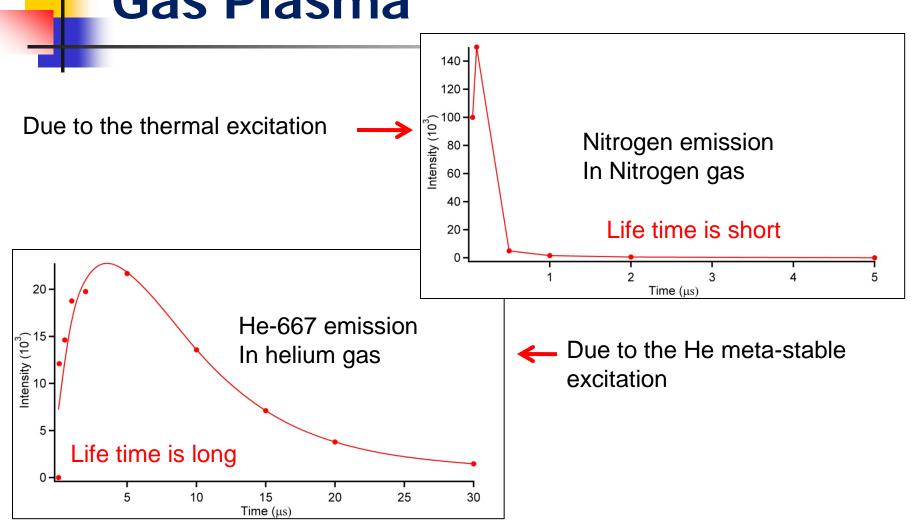
For gas Plasma: Nd-YAG (1064nm), 163mJ

For ablate sample: Nd-YAG (532nm), 74mJ

Delay Time between two laser: 10us

In this experiment, He gas plasma was made at 5mm in front the sample. After the gas plasma formation, second laser was irradiated for ablation with 10us delay time. Atoms come out from the sample and excited in the cooled He gas plasma.

Time Profile of Emission of the Gas Plasma



He emission proceeds in cooled plasma

Two possible mechanism

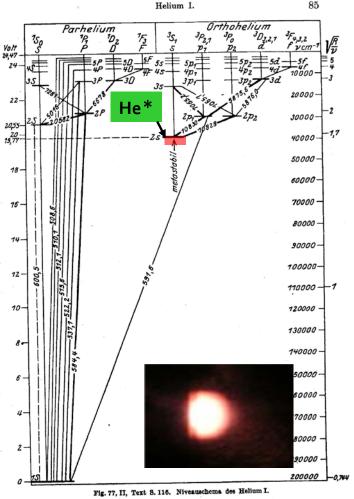
He meta - He meta collision

$$He^* + He^* \rightarrow He + (He^+ + e^-)$$

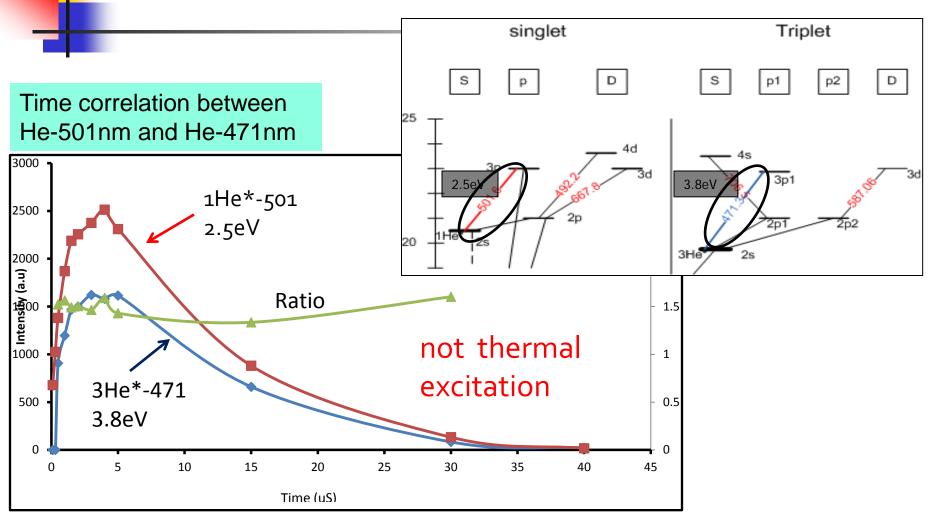
$$He^+ + e^- \rightarrow He^{**} \rightarrow He + hv$$

Thermal excitation from He meta

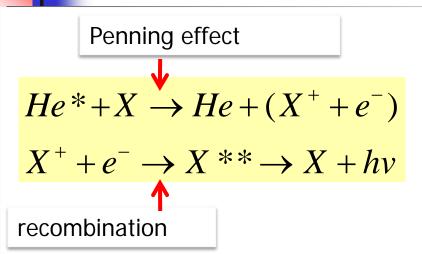
$$He^* + thermal \rightarrow He^{**} \rightarrow He + hv$$

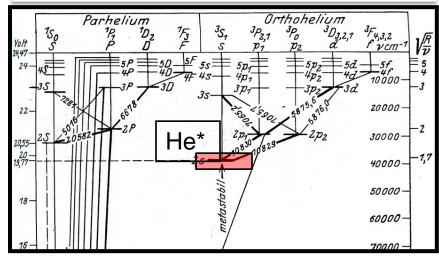


The evidence to prove He emission is not due to thermal excitation



The Excitation Mechanism through Helium Meta-stable





X atom collides with He metastable



X atom is ionized and Releases free electron

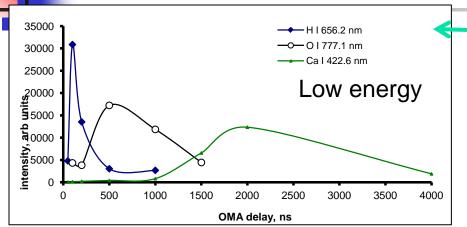


Electron recombines with X+ ion



Spectral emission of X atom

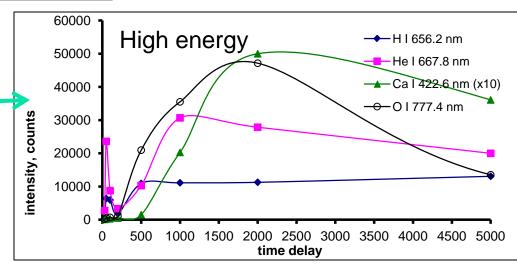
Time Profile of H, O, Ca emission in the cooled He gas plasma



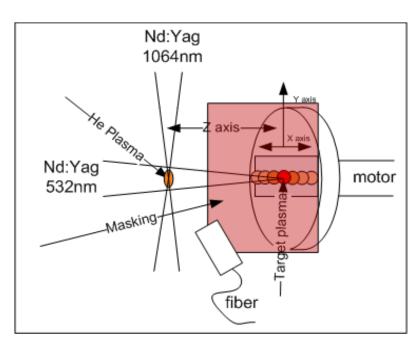
Due to Helium meta-stable excitation

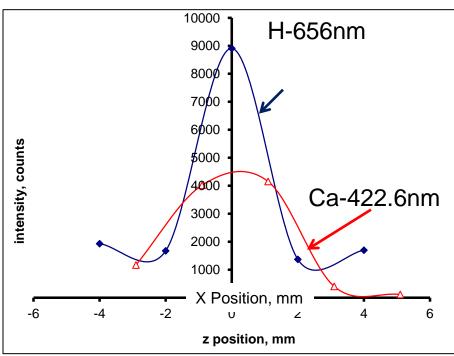
H atoms come out faster than other atoms, which proves the "mismatching effect"

Another mechanism (shock wave excitation) works for high energy pulse case



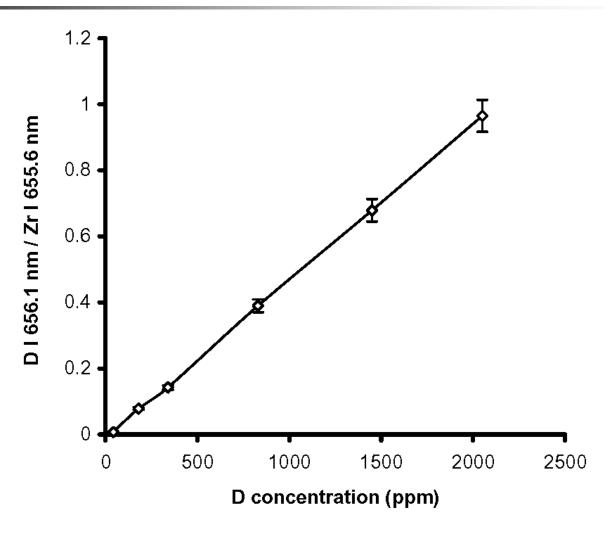
Direction of Gushing Atoms





Distribution of H atoms is more straight forwards compared to other atoms





Conclusion

 He emission takes place through the collision of two Helium meta-stable atoms (He*).

$$He^* + He^* \to He + (He^+ + e^-) \to He^{**} \to He^* + hv$$

 In a cooled He plasma, Helium meta-stable atoms (He*) excites atoms including Hydrogen

$$He^* + X \to He + (X^+ + e^-) \to He^{**} \to X + hv$$

 Hydrogen atoms gushed out faster than other atoms, which proved our hypothesis, namely "mismatching effect"

Published work

- Z.S. Lie, H. Niki, K. Kagawa, M.O. Tjia, R. Hedwig, M. Pardede, E. Jobiliong, M.M. Suliyanti, S.N. Abdulmadjid, K.H. Kurniawan, Observation of Exclusively He-Induced H Emission in Cooled Laser Plasma, J. Appl. Phys., 109, 10 (2011) pp. 103305 1-4
- R. Hedwig, Z.S. Lie, K.H. Kurniawan, A.N. Chumakov, K. Kagawa, M.O. Tjia, Toward Quantitative Deuterium Analysis with Laser-Induced Breakdown Spectroscopy Using Atmospheric-Pressure Helium Gas, J. Appl. Phys. 107, 2 (2010) pp. 023301 1-5
- Z.S. Lie, M. Pardede, R. Hedwig, M.M. Suliyanti, E. Steven, Maliki, Koo H. Kurniawan, M. Ramli, S.N. Abdulmadjid, N. Idris, K. Lahna, K. Kagawa and M.O. Tjia, Intensity Distributions of Enhanced H Emission from Laser-Induced Low-Pressure He Plasma and a Suggested He-Assisted Excitation Mechanism, J. Appl. Phys., 106, 3 (2009) pp. 043303 1-6
- Koo H. Kurniawan, T.J. Lie, M.M. Suliyanti, R. Hedwig, M. Pardede, M. Ramli, H. Niki, S.N. Abdulmadjid, N. Idris, K. Lahna, Y. Kusumoto, K. Kagawa, M.O. Tjia, The Role of He in Enhancing the Intensity and Lifetime of H and D Emissions from Laser-Induced Atmospheric-Pressure Plasma, J. Appl. Phys. 105 (2009) pp. 103303-1-6
- K.H. Kurniawan, M. Pardede, R. Hedwig, Z.S. Lie, T.J. Lie, D.P. Kurniawan, M. Ramli, K. Fukumoto, H. Niki, S.N. Abdulmadjid, N. Idris, T. Maruyama, K. Kagawa, M.O. Tjia, Quantitative Hydrogen Analysis of Zircaloy-4 Using Low-Pressure Laser Plasma Technique, Anal. Chem. 79,7 (2007) pp. 2703-2707
- M. Pardede, R. Hedwig, M.M. Suliyanti, Z.S. Lie, T.J. Lie, D.P. Kurniawan, K.H. Kurniawan, M. Ramli, K. Fukumoto, H. Niki, S.N. Abdulmadjid, N. Idris, T. Maruyama, K. Kagawa, M.O. Tjia, Comparative Study of Laser-Induced Plasma Emission of Hydrogen from Zircaloy-2 Samples in Atmospheric and Low Pressure Ambient Helium Gas, Appl. Phys. B. 89, 2-3 (2007) pp. 291 298.